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Physics Glencoe Physics: Principles and Problems, Student Edition The de Broglie wavelength and speed of an electron. The De-Broglie wavelength is  $= 7.77 \times 10^{-11} \text{ m}$  The speed of an electron is  $= 9.376 \times 10^6 \text{ m s}^{-1}$  Given data: The electron accelerated with a potential difference  $(V) = 250 \text{ V}$

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Formula used:  $KE = \frac{1}{2} m v^2 = e V$  Where, KE = kinetic energy m = mass e = charge of electron v = velocity V = potential difference Re-arrange the equation,  $v = \left( \frac{2 e V}{m} \right)^{1/2}$  is the required ...

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