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Alexandre V. Morozov 1 Solutions for Problem 1 a) $Q=0$ for adiabatic

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processes, and thus the first law of thermodynamics becomes: $U + A = 0$;
(1) where A is the work done by gas, and U is its internal energy. Using
 $A = P \Delta V$ and $U = C_V \Delta T$ we obtain: $C_V (T_2 - T_1) + P \Delta V = 0$;
(2)

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Thermal Physics

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