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activity series, predict and balance the
following single replacement reactions.
Use abbreviations to indicate the
appropriate phase of reactants and
products where possible. Note: Not all

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of the reactions will occur. For those that do not, write no reaction. l.

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Handbook Answers Answers for the questions in Chapter 8: ROUND 3
Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions. Use abbreviations to indicate the appropriate phase of reactants and products where possible.

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The Ultimate Chemical Equations

Handbook Determine the oxidation number of the underlined element:

KMnO₄ Since K is an alkali metal, its charge must be 1+ Oxygen is 2— but

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there are four of them, therefore, 4 times 2— equals 8— If 1+ and 8— are added Ultimate Chemical Equation Answers

The Ultimate Chemical Equations Handbook Answers Chapter 7
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